

Total number of printed pages-4

53 (ME 301) BATD

2019

BASIC THERMODYNAMICS

Paper : ME 301

Full Marks : 100

Time : Three hours

The figures in the margin indicate full marks for the questions.

Answer **any five** questions.

1. (a) Define : isothermal, isobaric, isochoric processes. 5
- (b) What are the parameters needed to be specified to define a thermodynamic system? 4
- (c) A system can be characterized by specifying P , V and T . Are they independent of one another? 3
- (d) Distinguish between intensive and extensive properties of a system. 3

Contd.

- (e) Distinguish between stored energy and energy in transit. What is the difference between gauge pressure and absolute pressure? $2.5 \times 2 = 5$
2. (a) What do you mean by positive and negative work interactions? 3
- (b) Distinguish between a path function and a point function. 3
- (c) Define work and what is its unit? 2
- (d) Give the relationship between the gauge, atmospheric and absolute pressure. 2
- (e) The pressure measured by a pressure gauge is 5000 kN/m^2 . Barometric pressure of the atmosphere is 760 mm of mercury. Find the absolute pressure. 5
- (f) What do you mean by pure substance? Give examples. 3
- (g) Explain briefly Zeroth law of thermodynamics. 2

3. (a) An artificial satellite revolves round the earth with a relative velocity of 800 m/s . If acceleration due to gravity is 9 m/s^2 and gravitational force is 3600 N , calculate its kinetic energy. 5

(b) What is PMM1? Justify with reasons whether it is feasible or not. 4

(c) A tube contains an oil of specific gravity 0.9 to a depth of 120 cm . Find the gauge pressure at this depth (in kN/m^2). 4

(d) To a closed system, 150 kJ of work is supplied. If the initial volume is 0.6 m^3 and pressure of the system changes as $P = 8 - 4V$, where (P) is in bar and (V) is in m^3 , determine the final volume and pressure of the system. 7

4. (a) Explain the following terms : $2 \times 5 = 10$

- (i) Triple point
 (ii) Latent heat of steam
 (iii) Dryness fraction of steam
 (iv) Superheated steam
 (v) Sensible heat of water.



- (b) A vessel having a volume of 0.6m^3 contains 3kg of liquid water and water vapour mixture in equilibrium at a pressure of 0.5MPa . Calculate :
- (i) Mass and volume of liquid
 - (ii) Mass and volume of vapour. 10
5. (a) Define heat engine, refrigerator and heat pump. 5
- (b) What is the perpetual motion machine of the second kind? 3
- (c) State the limitations of First law of thermodynamics. 3
- (d) Describe the working of a Rankine cycle. 9
6. (a) Compare external combustion and internal combustion engine. 5
- (b) What are the important basic components of an IC engine? Explain them briefly. 8
- (c) Write short notes on : $3.5 \times 2 = 7$
- (i) Thermodynamic equilibrium
 - (ii) Pressure measuring instruments.

