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53 (FPT 403) TPEN

2017

TRANSFER PROCESS ENGINEERING

Paper : FPT 403

Full Marks : 100

Time : Three hours

The figures in the margin indicate full marks for the questions.

Answer **any five** questions from **seven**.

1. (i) Write Newton's law of viscosity and define all terms. Discuss with diagram.
- (ii) Write Fourier's law of heat conduction.
- (iii) Calculate heat loss per m^2 of surface area for an insulating wall composed of 25.4mm thick fiber insulating board, where the inside temperature is 352.7K and the outside temperature is 297.1K. Thermal conductivity of insulating board is 0.048W/mK.

10+5+5=20

Contd.

2. (i) What is diffusive and convective mass transfer ?

(ii) Write Fick's 1st and 2nd law and define all terms.

(iii) Write unit of diffusion coefficient, mass transfer coefficient, mass flux, volumetric oxygen transfer coefficient.

8+8+4=20

3. Discuss briefly about "Interphase mass transfer" with diagram, mentioning required assumptions. 20

4. (i) How momentum flux and fluid linear velocity varies with radial distance ? Write the relation only, need not to derive.

(ii) A solute is diffused through a fluid at steady state. At point 1, solute concentration is $1.37 \times 10^{-2} \text{g/m}^3$ and at point 2, solute concentration is $0.72 \times 10^{-2} \text{g/m}^3$. The distance between the points is 0.4m . Diffusivity is $0.013 \text{m}^2/\text{s}$ and cross sectional area is constant. Calculate flux and concentration at the middle point of the distance.

8+12=20

5. (i) Write convective mass transfer equation and discuss.

(ii) Discuss any mass transfer model.

8+12=20

6. Discuss briefly mass transfer aspects in the following unit operations : 20

(i) Absorption

(ii) Adsorption

(iii) Distillation

(iv) Liquid-liquid extraction.

7. The equilibrium distribution of a solute in gas and liquid phase at constant temperature is given below :

$y = 1.2x$. Solute in bulk gas phase is 0.04 mole fraction and that in bulk liquid phase is 0.025. The individual mass transfer coefficient are $K_y = 7.2 \text{ Kmol/m}^2\text{h}(\Delta y)$ and

$K_x = 4.6 \text{ Kmol/m}^2\text{h}(\Delta y)$

Calculate :

(i) In which direction solute will transport ?

(ii) Calculate Flux.

(iii) Which phase resistance controls the mass transfer process here ?

20